

Refuge Denied The St Louis Passengers And The Holocaust

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riots and demonstrations the lifeblood of american social and political protest in the 1960s are now largely a historical memory but mary fainsod katzenstein argues that protest has not disappeared it has simply moved off the streets into the country s core institutions as a result conflicts over sexual harassment affirmative action and the rights of women gays and lesbians and people of color now touch us more than ever in our daily lives whether we are among those seeking change or those threatened by its

prospects no one is more aware of this than women demanding change from within the united states military and the american catholic church women in uniform are deeply patriotic and women active in the church are devoted to their callings yet katzenstein shows that these women often feel isolated and demeaned confronted by challenges as subtle as condescension and as blatant as career obstruction although faithful to their institutions many have proved fearless in their attempts to reshape them drawing on interviews with over a hundred women in the military and the church including senior officers combat pilots lay activists and nuns this book gives voice to the struggles and vision of these women as they have moved protest into the mainstream katzenstein shows why the military and the church similarly hierarchical and insistent on obedience have come to harbor deeply different forms of protest she demonstrates that women in the military have turned to the courts and congress whereas feminists in the church have used discursive protests writing organizing workshops and conferences to rethink in radical ways the meanings of faith and justice these different strategies she argues reflect how the law regulates the military but leaves the church alone faithful and fearless calls our attention to protest within institutions as a new stage in the history both of feminism and of social movements in america the book is an inspiring account of strength in the face of adversity and a groundbreaking contribution to the study of american feminism social protest and the historical development of institutions in american society

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with appendices

this book was written over a period of some 18 years to combat numerous modern misconceptions about the new testament its history its authors both as given by early christians and pagans the reasons it was considered inspired and the meaning from the actual wording of the new testament modern scholars are so wrong in their assessments of the new testament with their seemingly logical proofs but upon deeper examination those proofs are seen as false assumptions without any real proof and contradicted by the new testament itself and its early history from the beginnings of christianity

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the essays in the other missouri history explore a wide range of topics in missouri social history by dealing with the lives of ordinary missourians these pieces examine the effects of significant social and economic change at all levels of society with a broader scope in missouri history than previous studies this book demonstrates how missourians have been affected by issues of race class and gender gregg andrews s essay the racial politics of reconstruction in ralls county 1865 1870 examines how race shaped the political culture in ralls county during the reconstruction era andrews argues that race baiting was used prominently by editors of the ralls county record to discredit radicals in the county and was perhaps the most powerful political weapon that conservatives and later democrats could use to gain the allegiance of voters farmers are another popular topic for those practicing the other missouri history michael j steiner s the failure of alliance populism in northern missouri provides insight into the economic and rhetorical reasons for the failure of populism in missouri steiner contends that white farmers in northern missouri were happy with the status quo and rejected calls for radical reform and major change in the agricultural economy women began to become active in public life during the late nineteenth and early twentieth centuries janice brandon falcone s constance runcie and the runcie club of st joseph examines the first two decades of an important women s club that still exists in st joseph missouri also included in the other missouri history are essays by deborah j henry daniel a graff bonnie stepenoff robert faust and amber r clifford because of the diverse issues addressed this volume will appeal to general readers of missouri and midwestern history as well as to those who teach courses in history and have sought a supplemental text

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At What Temperature Does Water Boil? A Simple Explanation

Water, the elixir of life, is ubiquitous and essential. Understanding its properties, especially its boiling point, is crucial for various aspects of our daily lives, from cooking to scientific experiments. This article will demystify the seemingly simple question: "At what temperature does water boil?" and explore the factors that influence this crucial temperature.

The Basics: What is Boiling?

Boiling isn't just about water getting hot; it's a phase transition. When water boils, it changes from a liquid to a gas (water vapor or steam). This happens when the water molecules gain enough energy to overcome the attractive forces holding them together in the liquid state. They escape the liquid's surface, forming bubbles that rise to the top. This energy input is provided by heat. The temperature at which this occurs is called the boiling point.

Standard Boiling Point: 100°C (212°F)

Under standard atmospheric pressure (1 atmosphere or 101.325 kPa, which is the average air pressure at sea level), water boils at 100 degrees Celsius (100°C) or 212 degrees Fahrenheit (212°F). This is a widely accepted value, used as a benchmark in numerous applications. Imagine cooking pasta. You bring the water to a rolling boil, meaning the water temperature has reached 100°C. The bubbles you see are the escaping water vapor. This temperature is essential for cooking the pasta properly.

The Influence of Atmospheric Pressure: Higher Up, Lower Down

The key factor influencing water's boiling point is atmospheric pressure. Atmospheric pressure is the weight of the air above a given location. The higher the altitude, the less air there is above you, resulting in lower atmospheric pressure. At higher altitudes, like mountain tops, the atmospheric pressure is lower. This means that the water molecules require less energy to overcome the reduced pressure and escape into the gaseous phase. Consequently, water boils at a lower temperature. For instance, at the summit of Mount Everest, water boils at approximately 70°C (158°F). Conversely, at lower altitudes or under increased pressure (like in a pressure cooker), the atmospheric pressure is higher. Water molecules need more energy to overcome the increased pressure, causing the boiling point to increase. A pressure cooker utilizes this principle to cook food faster at higher temperatures.

Impurities and Boiling Point: A Minor Effect

While atmospheric pressure has a significant effect, impurities in the water have a relatively minor impact on the boiling point. Dissolved salts or other substances can slightly elevate the boiling point, a phenomenon known as boiling point elevation. However, this change is usually insignificant for everyday purposes. The increase is typically a fraction of a degree Celsius. Think about adding salt to boiling water while cooking. While it slightly increases the boiling point, the difference is negligible in most cooking scenarios.

Other Factors Affecting Boiling: Nucleation Sites

Another often overlooked factor is the presence of nucleation sites. These are tiny imperfections or irregularities on the surface of the container where bubbles can form

more easily. Without sufficient nucleation sites, water can become superheated – reaching temperatures significantly above its boiling point without actually boiling. This can be dangerous, leading to sudden, violent boiling. A slightly rough surface or adding boiling chips (small, inert porous materials) can provide sufficient nucleation sites.

Actionable Takeaways and Key Insights

Water boils at 100°C (212°F) at standard atmospheric pressure (sea level). Atmospheric pressure significantly impacts boiling point: lower pressure means a lower boiling point, and vice versa. Impurities have a minimal effect on the boiling point in most everyday situations. Nucleation sites are crucial for controlled boiling; a lack of them can lead to superheating. Understanding the boiling point of water is essential for various applications, from cooking to industrial processes.

Frequently Asked Questions (FAQs)

1. Can water boil at room temperature? Yes, but only under significantly reduced pressure. This is achieved in a vacuum chamber. 2. Why does salt make water boil faster? Salt doesn't actually make the water boil faster; it slightly raises the boiling point. The perceived faster boil is often due to other factors, such as increased heat transfer. 3. What happens if you keep boiling water after it reaches 100°C? The water will continue to boil, but its temperature will remain at 100°C (at standard pressure) as the added heat energy is used to convert liquid water into steam. 4. Does the amount of water affect its boiling point? No, the amount of water does not affect its boiling point. The boiling point depends solely on pressure and minor impurity effects. 5. Is it safe to drink water that has been boiled? Yes, boiling water kills most harmful bacteria and viruses, making it safe to drink. However, ensure the water source is initially safe.

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